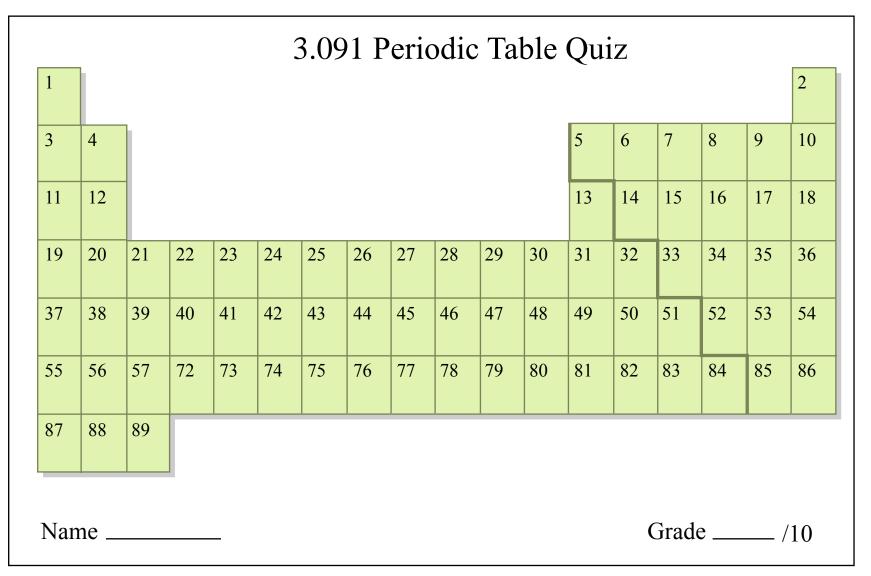
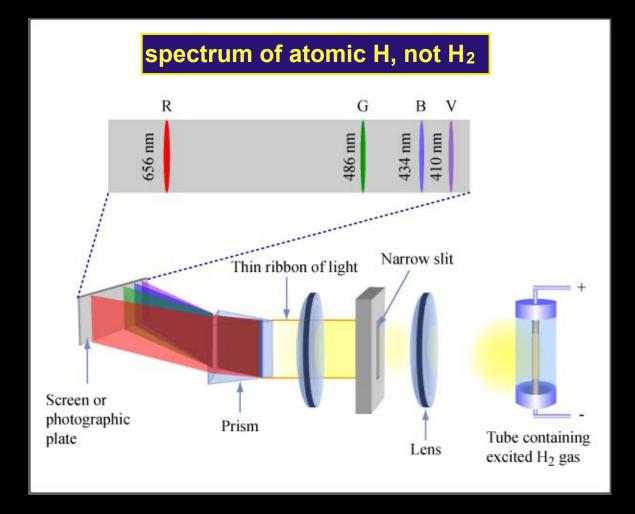
Welcome to 3.091

Lecture 5 September 18, 2009

Electron Shell Model & Quantum Numbers



Prism Spectrograph A.A. Ångström (1853)



Electronic emission transition

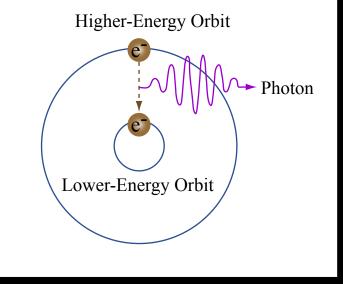


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Bohr Postulates for the Hydrogen Atom

- 1. Rutherford atom is correct
- 2. Classical EM theory not applicable to orbiting e
- 3. Newtonian mechanics applicable to orbiting e-

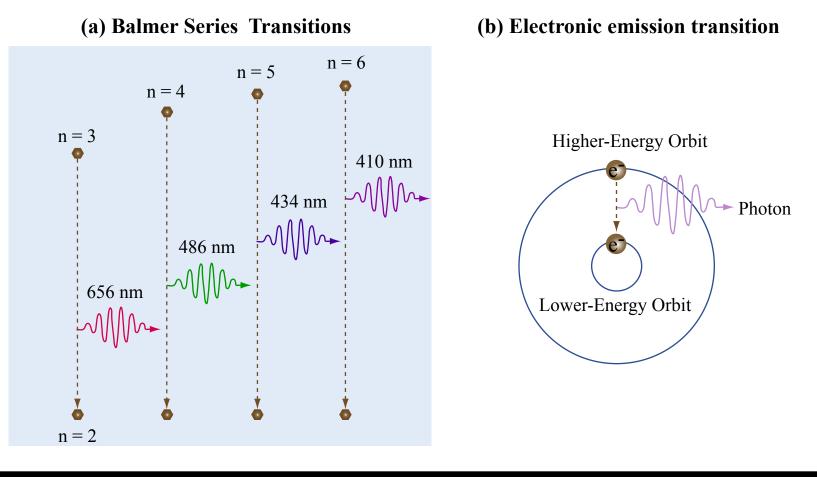
4.
$$E_{\text{electron}} = E_{\text{kinetic}} + E_{\text{potential}}$$

5. e⁻ energy quantized through its angular momentum:

$$L = mvr = nh/2\pi, n = 1, 2, 3,...$$

6. Planck-Einstein relation applies to e⁻ transitions:

$$\Delta E = E_{\rm f} - E_{\rm i} = h\nu = hc/\lambda$$
$$c = \nu\lambda$$







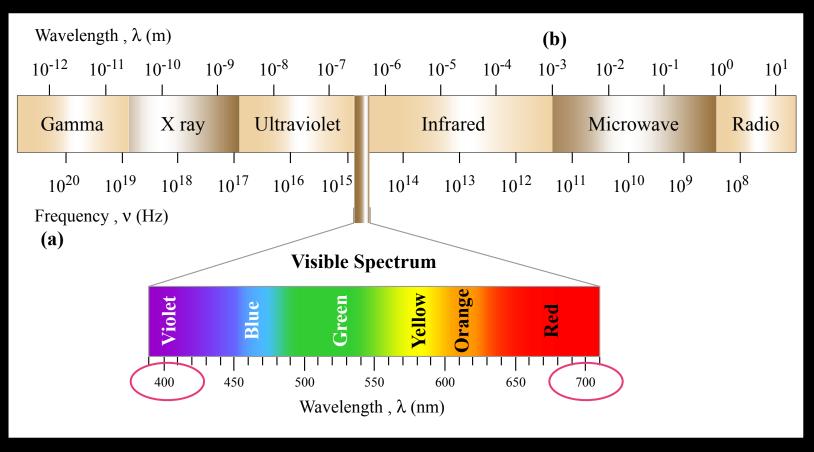


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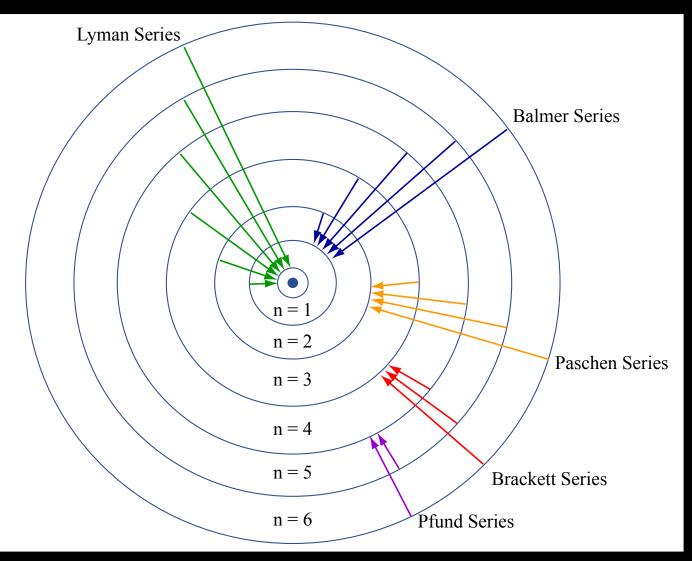


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"for their discovery of the laws governing the impact of an electron upon an atom"

Gustav Ludwig Hertz

• 1/2 of the prize Germany

Halle University Halle, Germany

B. 1887

D. 1975

James Franck

● 1/2 of the prize Germany

Goettingen University Goettingen, Germany

- B. 1882
- D. 1964

Prism Spectrograph A.A. Ångström (1853)

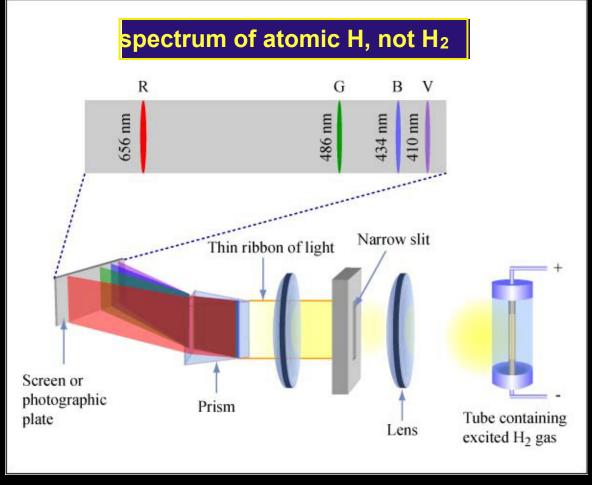


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"for his optical precision instruments and the spectroscopic and metrological investigations carried out with their aid"

Albert Abraham Michelson

USA

University of Chicago Chicago, IL, USA b. 1852 (in Strelno, then Germany) d. 1931



"in recognition of the extraordinary service they rendered by their researches into the influence of magnetism upon radiation phenomena"

Hendrik Antoon Lorentz

1/2 of the prize

the Netherlands

Leiden University Leiden, the Netherlands

b. 1853d. 1928

Pieter Zeeman

1/2 of the prize

the Netherlands

Amsterdam University Amsterdam, the Netherlands

b. 1865 d. 1943

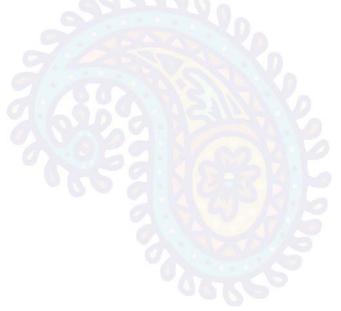
"for his discovery of the Doppler effect in canal rays and the splitting of spectral lines in electric fields"

Johannes Stark

Germany

Greifswald University Greifswald, Germany

b. 1874d. 1957



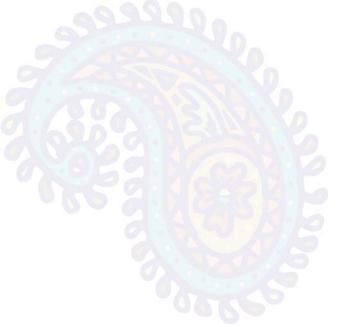
"for his services in the investigation of the structure of atoms and of the radiation emanating from them"

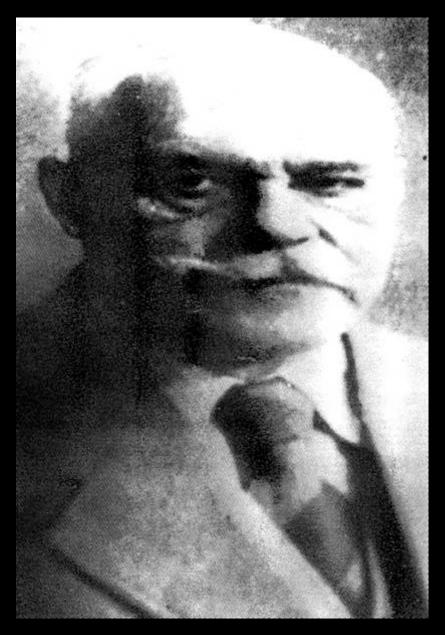
Niels Henrik David Bohr

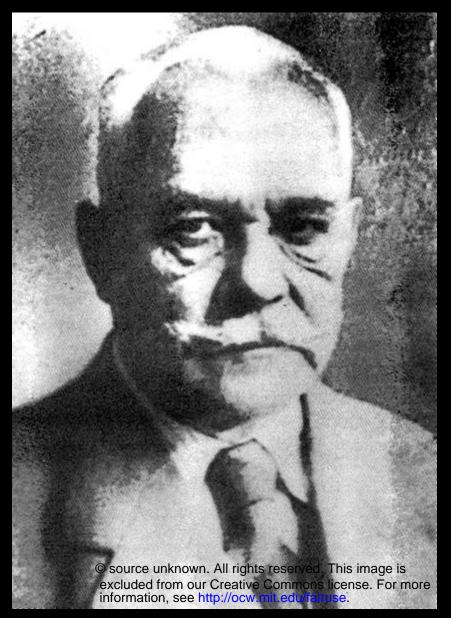
Denmark

Copenhagen University Copenhagen, Denmark

b. 1885d. 1962







Arnold Sommerfeld on the occasion of his 80th birthday

Schematic representation of the Stern Gerlach experiment (1921).

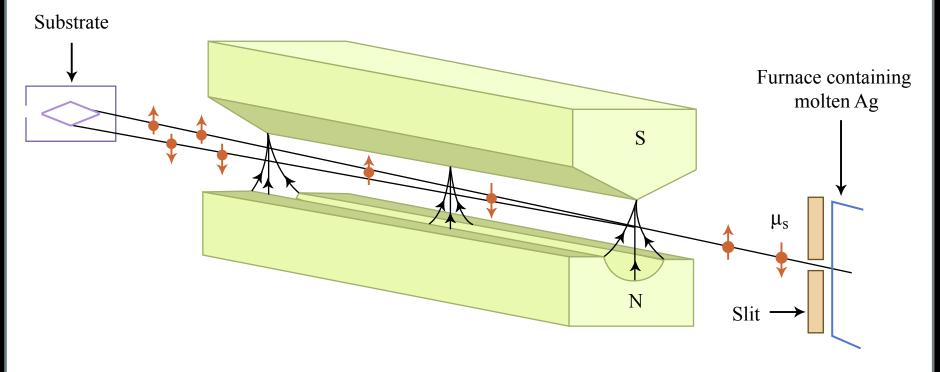
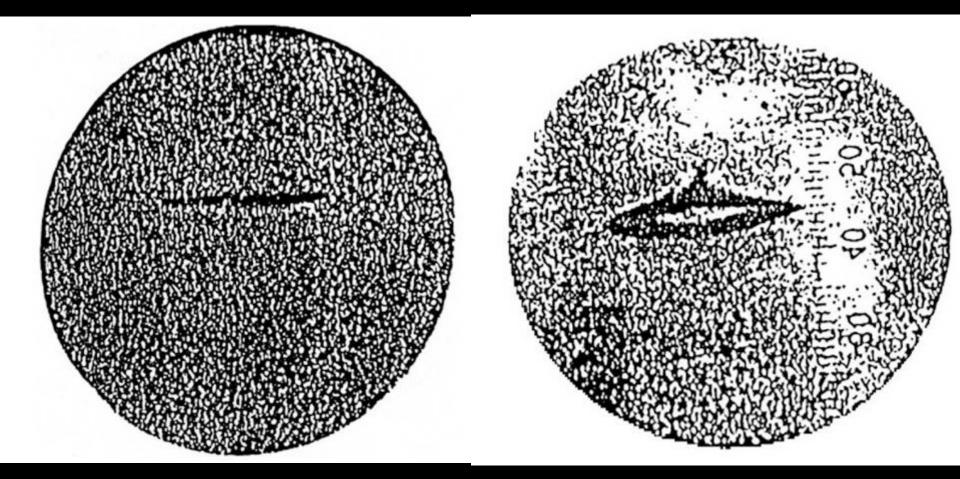


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no magnetic field divergent magnetic field

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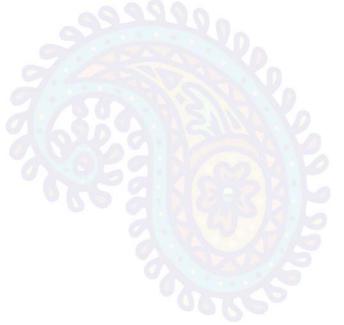
"for his contribution to the development of the molecular ray method and his discovery of the magnetic moment of the proton"

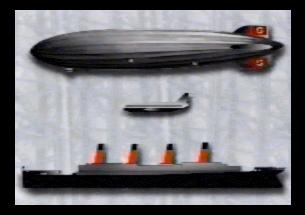
Otto Stern

USA

Carnegie Institute of Technology Pittsburgh, PA, USA

b. 1888 (in Sorau, then Germany) d. 1969





Hindenburg

B 747

Titanic

HYDROGEN IN TRANSPORTATION

* built in Germany by Zeppelin (LZ129)

* 135 ft in dia. \times 804 ft long

- * 7 million cu ft gas
- * 112 tons useful lift

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* Helium Control Act 🖙 inflated with H₂

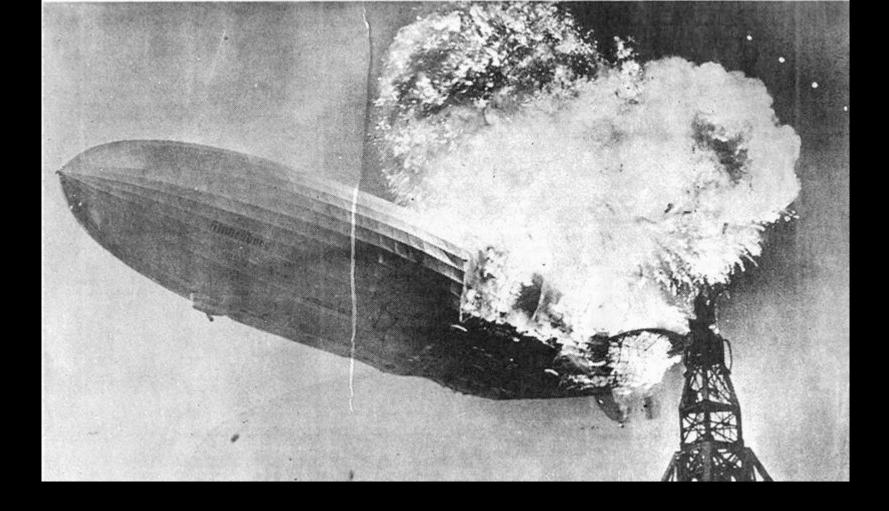
Hindenburg



* 10 transatlantic flights in 1936: 1002 passengers

- * cruising speed 78 mph
- * May 6, 1937 arrival of first flight to US

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* while docking at Lakehurst, NJ

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COLUMN STATE

* 35 dead of 97 on board

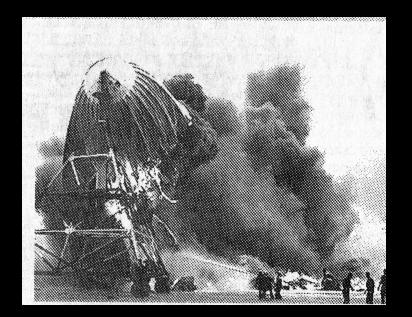
* electrical discharge in vicinity of H₂ leak

* skin of resin (finished with lacquer dope)

 + aluminum powder
 + iron oxide on inside

Sheets of SOLID ROCKET MOTOR grains

* end of rigid airships in commercial air transportation



* US Navy airship filled with helium (1956)

* two-ply cotton envelope could not withstand gasoline fire

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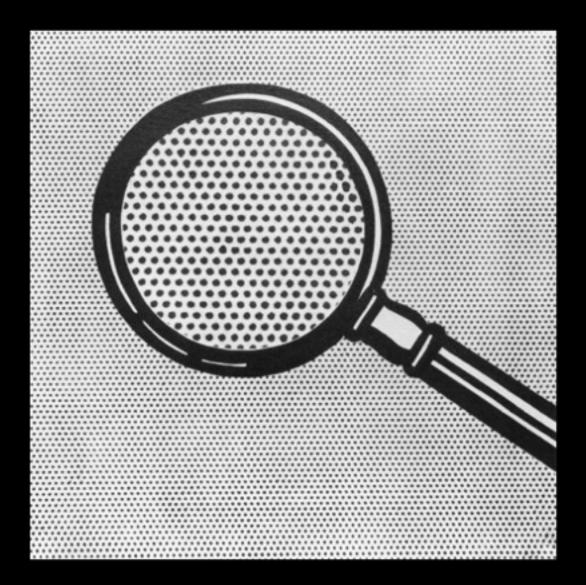


Photo courtesy of action datsun on Flickr.

Vicki! I - I Thought I Heard Your Voice!

Roy Lichtenstein (1964)

enamel on steel



Magnifying Glass

Roy Lichtenstein (1963)

oil on canvas

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3.091SC Introduction to Solid State Chemistry Fall 2009

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