## WAVE-PARTICLE DUALITY of LIGHT and MATTER

## (A) Light (electromagnetic radiation)

## Light as a wave

For now we neglect the polarization vector orientation Propagating in x-direction:

$$\xi_{0} = \int_{0}^{\infty} \int_{0}^$$

Superposition principle





(out of phase by  $\lambda/2$ )

This leads to many interference phenomena

Young's 2-slit experiment





If just a wave, expect light to scatter off electron



The backscattered wave is red-shifted  $(\lambda' > \lambda)$ , i.e. less energy/photon.

$$E' = \frac{hc}{\lambda'} < \frac{hc}{\lambda} = E$$

Energy (and momentum) transferred to the electron. Need relativistic mechanics to solve

$$p = \frac{hv}{c} \left(=\frac{h}{\lambda}\right) \quad \text{for the light}$$

Light is a particle with energy

momentum 
$$p = \frac{hv}{c}$$

Light can behave <u>both</u> as a wave <u>and</u> as a particle!! Which aspect is observed depends on what is measured.

## (B) <u>Matter</u>

e-

<u>Matter as particles</u>  $\Rightarrow$  obvious from everyday experience

Matter as waves (deBroglie, 1929, Nobel Prize for his Ph.D. thesis!)

Same relationship between momentum and wavelength for light and for matter

$$p = \frac{h}{\lambda} \implies \lambda = \frac{h}{p} \equiv de Broglie wavelength$$

Amazing notion! But wavelength only observable for microscopic momentum





Consequences (I)

(1) on Bohr atom



If e- wave does not close on itself, eventually destructive interference will kill it!



I f e- wave does close on itself, then constructive interference preserves it.

Criterion for stability:

$$2\pi r = n\lambda = \frac{nh}{p} = \frac{nh}{mv}$$
  
or  $mvr = \frac{nh}{2\pi} = n\hbar$   
 $\Rightarrow \ell = n\hbar$ 

As Bohr had assumed angular momentum is quantized!