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3.23 Electrical, Optical, and Magnetic Properties of Materials
Fall 2007

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3.23 Fall 2007 – Lecture 7

ONE BLOCH AT A TIME

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Last time

1. Vector space (expectation values measure the projection on different eigenvectors)
2. Eigenvalues and eigenstates as a linear algebra problem
3. Variational principle
4. Its application to a H atom (atomic units)
5. Hamiltonian for a molecular system; bonding and antibonding states
6. Potential energy surface of a molecule
7. Vibrations at equilibrium; quantum harmonic oscillator

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Study

- Chapter 2 of Singleton textbook – “Band theory and electronic properties of solids”

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Dynamics, Lagrangian style

- First construct $L=T-V$
- Then, the equations of motion are given by

$$\frac{d}{dt} \left(\frac{\partial L}{\partial \dot{q}_j} \right) - \frac{\partial L}{\partial q_j} = 0$$

(the dot is a time derivative)

- Why ? We can use generalized coordinates. Also, we only need to think at the two scalar functions T and V

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Newton's second law, too

- 1-d, 1 particle: $T=1/2 mv^2$, $V=V(x)$

$$\frac{d}{dt} \left(\frac{\partial L}{\partial \dot{q}_j} \right) - \frac{\partial L}{\partial q_j} = 0$$

$$\frac{d}{dt} \left(\frac{\partial \left(\frac{1}{2} m \dot{x}^2 \right)}{\partial \dot{x}} \right) + \frac{\partial V}{\partial x} = 0 \quad \Rightarrow \quad \frac{d}{dt} (m\dot{x}) = - \frac{\partial V}{\partial x}$$

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Hamiltonian

- We could use it to derive Hamiltonian dynamics (twice the number of differential equations, but all first order). We introduce a Legendre transformation

CONJUGATE
VAR. \rightarrow

$$p_i = \frac{\partial L}{\partial \dot{q}_i} \quad H(q, p, t) = \sum_i \dot{q}_i p_i - L(q, \dot{q}, t)$$

$$\dot{q}_i = \frac{\partial H}{\partial p_i} \quad - \dot{p}_i = \frac{\partial H}{\partial q_i}$$

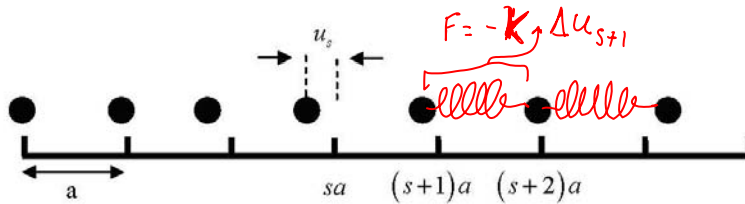
$E(V, S) \rightarrow$

$H = E + PV =$
 $= H(P, S)$

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$$L = T - V$$

1-dimensional monoatomic chain



$$H = T + V = \sum_s \frac{p_s^2}{2m} + \frac{1}{2} K \sum (u_s - u_{s+1})^2$$

$$q_i = \frac{\partial H}{\partial p_i} \quad -\dot{p}_i = \frac{\partial H}{\partial q_i}$$

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$$H = \sum_s H_s \quad H_s = \frac{p_s^2}{2m} + \frac{1}{2} K (u_s - u_{s+1})^2$$

$$M \frac{d^2 u_s}{dt^2} = K (u_{s+1} - u_{s-1} - 2u_s)$$

ANFATZ

$$u_s \propto e^{-i\omega t}$$

$$-M\omega^2 u_s = K (u_{s+1} + u_{s-1} - 2u_s)$$

$$u_s = u e^{i s k a}$$

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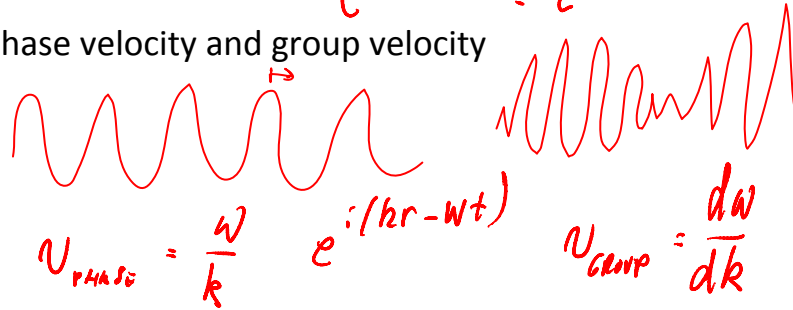
Properties

- Unique solutions for k in the first BZ

$$\frac{u_s}{u_{s+1}} = \frac{e^{ihsa}}{e^{-i(h+\frac{2\pi}{a})a}} = e^{-ika} \quad k \rightarrow k + \frac{2\pi}{a}n$$

$$e^{-i(h+\frac{2\pi}{a})a} = e^{-ika}$$

- Phase velocity and group velocity



$$v_{\text{phase}} = \frac{\omega}{k} \quad e^{i(kr - \omega t)} \quad v_{\text{group}} = \frac{d\omega}{dk}$$

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Properties

- Standing waves

$$\frac{d\omega}{dk} = 0 \quad k = \pm \frac{\pi}{a}$$

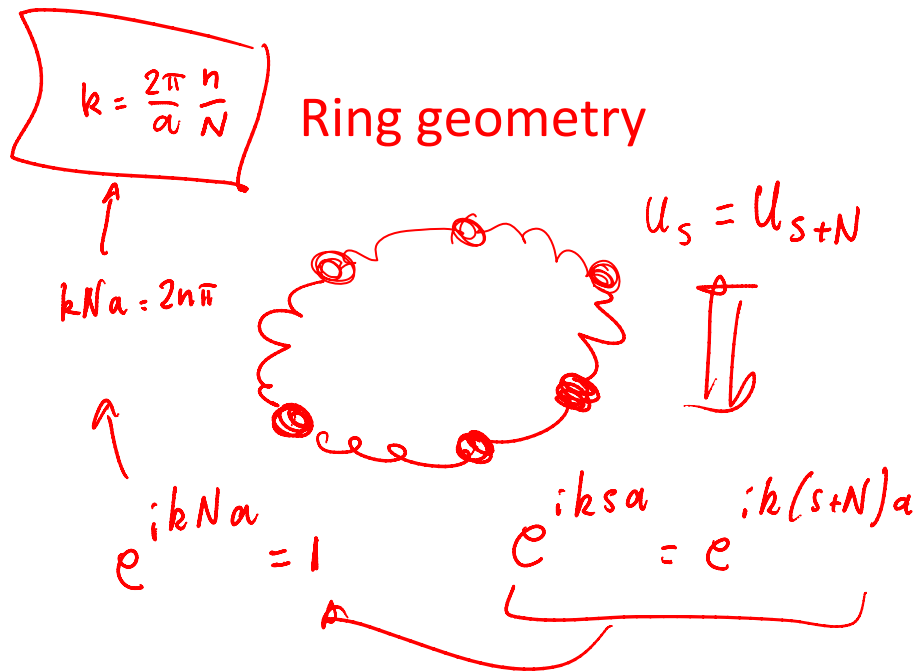
- Long wavelength limit

$$\omega^2(k) = \frac{2K}{M} (1 - \cos ka)$$

$$ka \ll 1 \quad 1 - \frac{(ka)^2}{2}$$

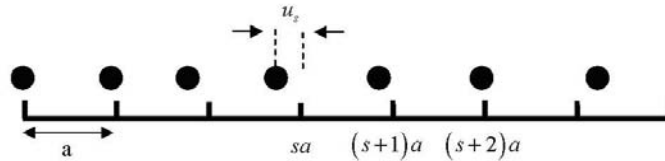
$$\omega^2 = \frac{2K}{M} \frac{(ka)^2}{2} \Rightarrow \omega \left(\frac{k}{M} \right)^{\frac{1}{2}} ka$$

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1-dimensional diatomic chain



III. Equations of motion

$$M \frac{d^2 u_{1s}}{dt^2} = K(u_{2s} - u_{1s}) + G(u_{2s-1} - u_{1s})$$

$$M \frac{d^2 u_{2s}}{dt^2} = K(u_{1s} - u_{2s}) + G(u_{1s+1} - u_{2s})$$

IV. Solutions

$$u_{1s} = u_1 e^{iks} e^{-i\omega t}, \quad u_{2s} = u_2 e^{iks} e^{-i\omega t}$$

V. Dispersion relations

$$(M\omega^2 - (K + G))u_1 + (K + Ge^{-i\omega a})u_2 = 0$$

$$(K + Ge^{i\omega a})u_1 + (M\omega^2 - (K + G))u_2 = 0$$

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The homogenous linear equations have a solution only if the determinant of the coefficients is zero:

$$\begin{vmatrix} (M\omega^2 - (K+G)) & (K + Ge^{-ika}) \\ (K + Ge^{ika}) & (M\omega^2 - (K+G)) \end{vmatrix} = 0$$

with solutions:

$$\omega^2 = \frac{K+G}{M} \pm \frac{1}{M} \sqrt{K^2 + G^2 + 2KG \cos ka}$$

$$\frac{u_1}{u_2} = \mp \frac{K + Ge^{-ika}}{K + Ge^{ika}}$$

for each k there are two solutions which are called the two branches of the dispersion curves.

Image removed due to copyright restrictions.

Please see Fig. 22.10 in Ashcroft, Neil W., and N. David Mermin. *Solid State Physics*. Belmont, CA: Brooks/Cole, 1976. ISBN: 9780030839931.

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Translational Symmetry

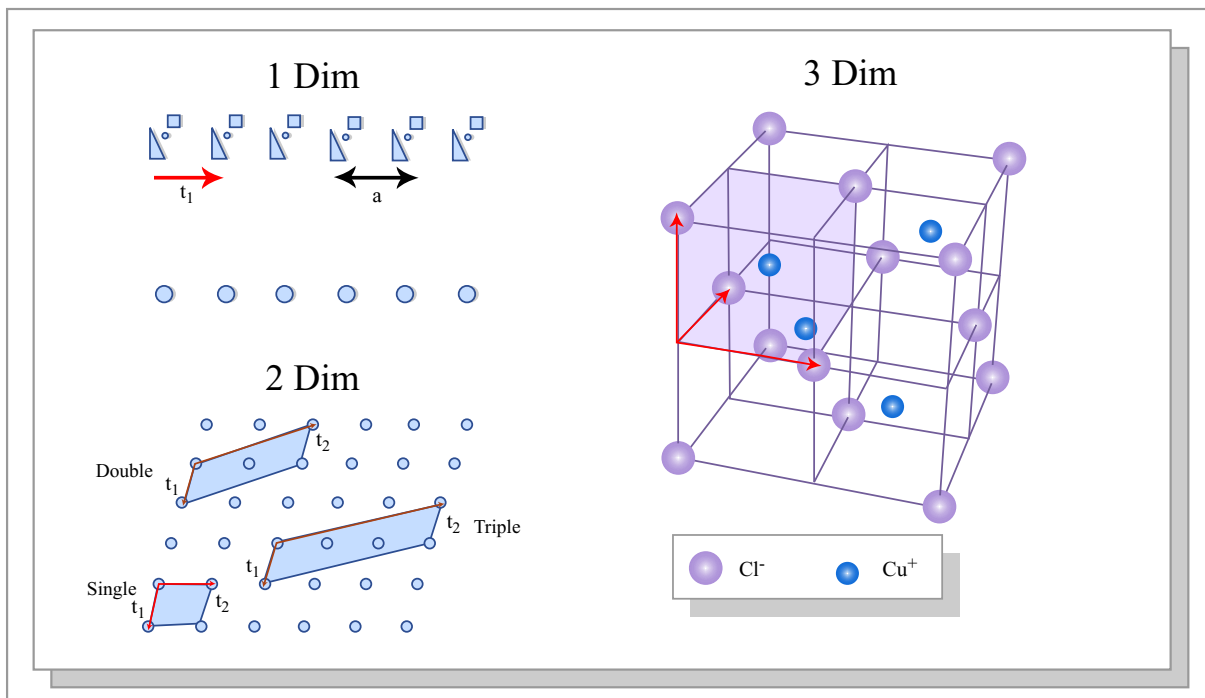


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Bravais Lattices

- Infinite array of points with an arrangement and orientation that appears exactly the same regardless of the point from which the array is viewed.

$$\vec{R} = l\vec{a}_1 + m\vec{a}_2 + n\vec{a}_3 \quad l, m \text{ and } n \text{ integers}$$

\vec{a}_1, \vec{a}_2 and \vec{a}_3 primitive lattice vectors

- 14 Bravais lattices exist in 3 dimensions (1848)
- M. L. Frankenheimer in 1842 thought they were 15. So, so naïve...















Bravais Lattice	Parameters	4 Lattice Types			
		Simple (P)	Volume Centered (I)	Base Centered (C)	Face Centered (F)
Triclinic	$a_1 \neq a_2 \neq a_3$ $a_{12} \neq a_{23} \neq a_{31}$				
Monoclinic	$a_1 \neq a_2 \neq a_3$ $a_{23} = a_{31} = 90^\circ$ $a_{12} \neq 90^\circ$				
Orthorhombic	$a_1 \neq a_2 \neq a_3$ $a_{12} = a_{23} = a_{31} = 90^\circ$				
Tetragonal	$a_1 = a_2 \neq a_3$ $a_{12} = a_{23} = a_{31} = 90^\circ$				
Trigonal	$a_1 = a_2 = a_3$ $a_{12} = a_{23} = a_{31} < 120^\circ$				
Cubic	$a_1 = a_2 = a_3$ $a_{12} = a_{23} = a_{31} = 90^\circ$				
Hexagonal	$a_1 = a_2 \neq a_3$ $a_{12} = 120^\circ$ $a_{23} = a_{31} = 90^\circ$				

Figure by MIT OpenCourseWare.

Symmetry

- Symmetry operations: actions that transform an object into a new but undistinguishable configuration
- Symmetry elements: geometric entities (axes, planes, points...) around which we carry out the symmetry operations

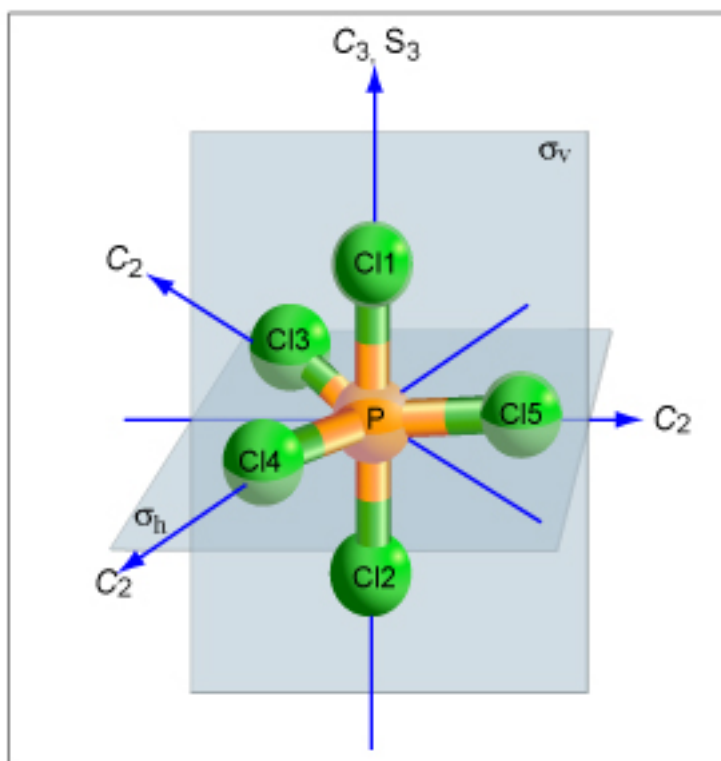


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Symmetry elements and their corresponding operations

Symmetry elements		Symmetry operations	
E	Identity	E	leave molecule unchanged
C_n	n-Fold rotation axis	$\hat{C}_n, \hat{C}_n^2, \dots, \hat{C}_n^n$	rotate about axis by $360^\circ/n$ 1, 2, ..., n times (indicated by superscript)
σ	Mirror plane	$\hat{\sigma}$	reflect through the mirror plane
i	Inversion center	\hat{i}	$(x, y, z) \rightarrow (-x, -y, -z)$
S_n	n-Fold rotation-reflection axis	\hat{S}_n	rotate about axis by $360^\circ/n$, and reflect through a plane perpendicular to axis.

Figure by MIT OpenCourseWare.

Group Therapy...

A group G is a finite or infinite set of elements A, B, C, D...together with an operation " \odot " that satisfy the four properties of:

- Closure:** If A and B are two elements in G, then $A \odot B$ is also in G.
- Associativity:** For all elements in G, $(A \odot B) \odot C = A \odot (B \odot C)$.
- Identity:** There is an identity element I such that $I \odot A = A \odot I = A$ for every element A in G.
- Inverse:** There is an inverse or reciprocal of each element. Therefore, the set must contain an element $B = \text{inv}(A)$ such that $A \odot \text{inv}(A) = \text{inv}(A) \odot A = I$ for each element of G.

Examples

- Integer numbers, and addition
- Integer numbers, and multiplication
- Real numbers, and multiplication
- Rotations around an axis by $360/n$

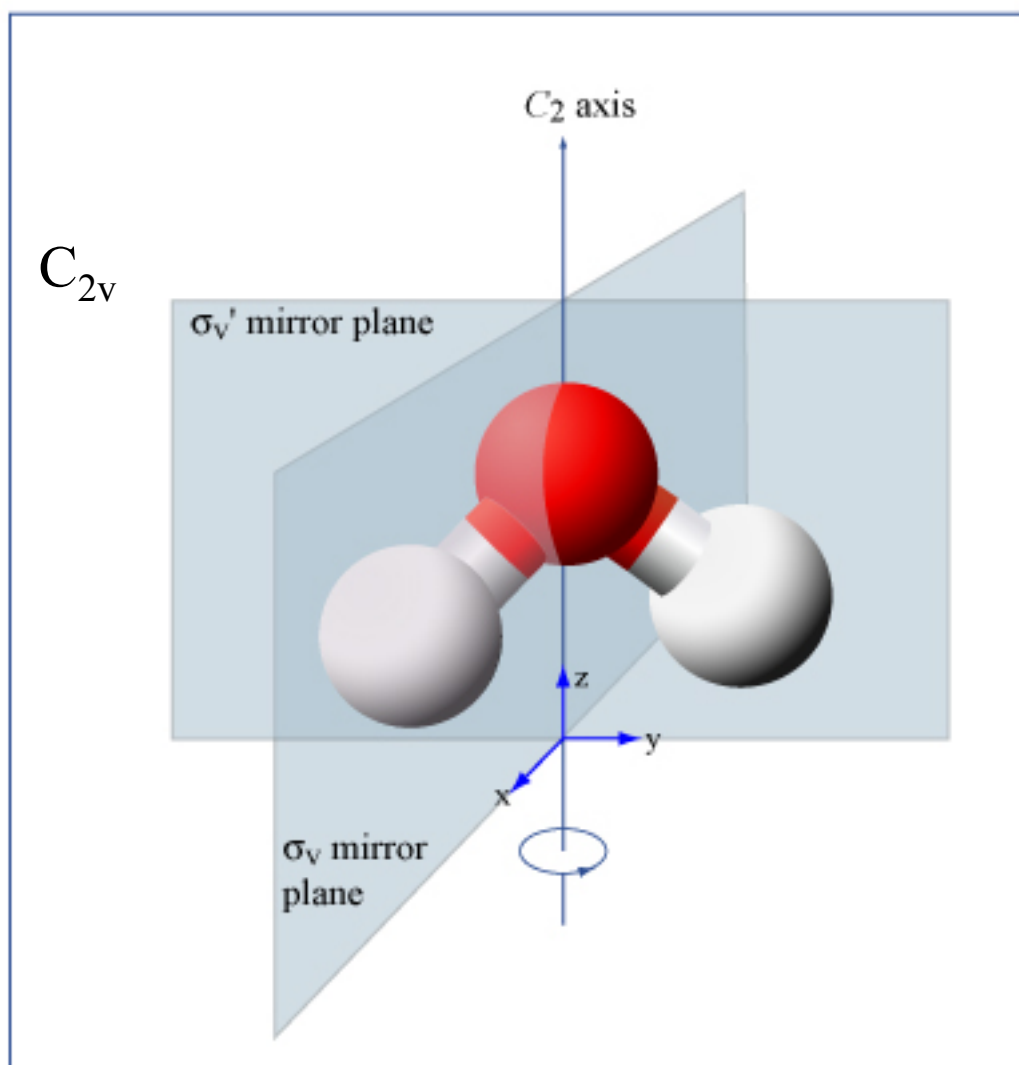


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Symmetries of H₂O

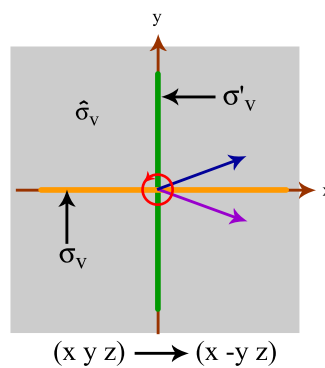
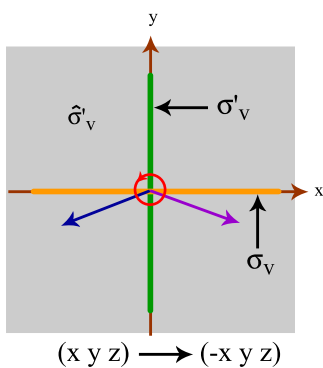
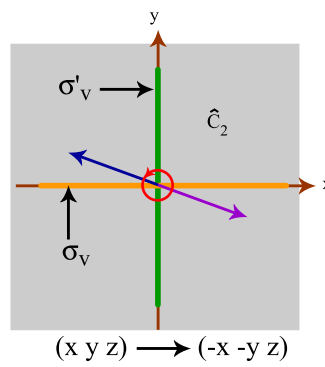
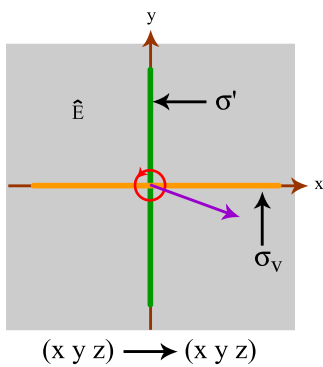
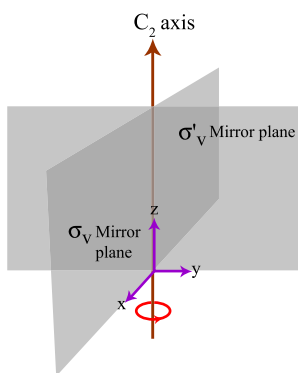


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Symmetries of H₂O

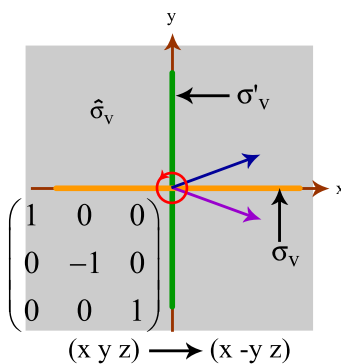
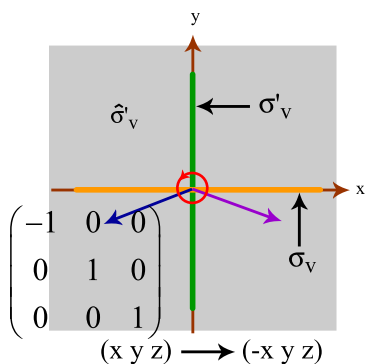
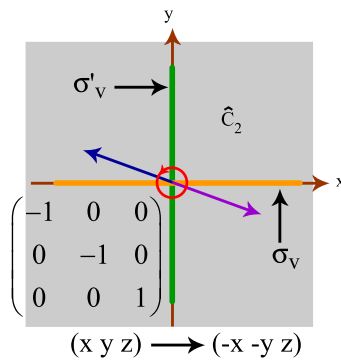
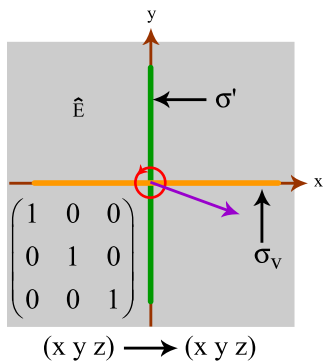
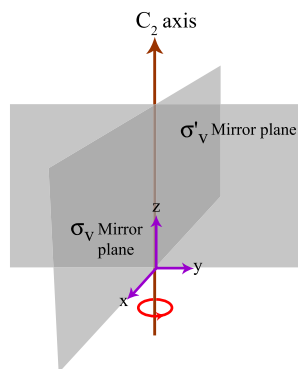


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The 4 symmetry operations of H₂O form a group (called C_{2v})


1. **Closure:** A⊗B is also in G.
2. **Associativity:** (A⊗B)⊗C=A⊗(B⊗C)
3. **Identity:** I⊗A=A⊗I
4. **Inverse:** A⊗inv(A)=inv(A)⊗A=I


Second Operation	First Operation			
	\hat{E}	\hat{C}_2	$\hat{\sigma}_v$	$\hat{\sigma}'_v$
\hat{E}	\hat{E}	\hat{C}_2	$\hat{\sigma}_v$	$\hat{\sigma}'_v$
\hat{C}_2	\hat{C}_2	\hat{E}	$\hat{\sigma}'_v$	$\hat{\sigma}_v$
$\hat{\sigma}_v$	$\hat{\sigma}_v$	$\hat{\sigma}'_v$	\hat{E}	\hat{C}_2
$\hat{\sigma}'_v$	$\hat{\sigma}'_v$	$\hat{\sigma}_v$	\hat{C}_2	\hat{E}


Figure by MIT OpenCourseWare.


Ten crystallographic point groups in 2d


1
 C₁

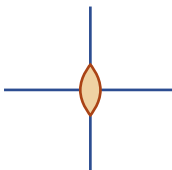
2
 C₂

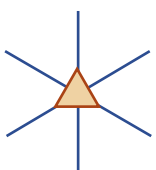
3
 C₃

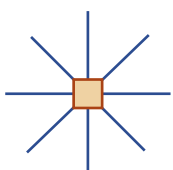
4
 C₄

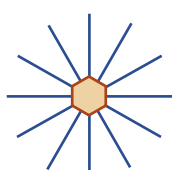
6
 C₆

m
 C_s

2mm
 C_{2v}

3m
 C_{3v}

4mm
 C_{4v}

6mm
 C_{6v}

The ten crystallographic plan point groups. Upper symbol, international notation; lower symbol, Schoenflies notation (see text).

Figure by MIT OpenCourseWare.

The Crystallographic Point Groups and the Lattice Types.

Crystal System	Schoenflies Symbol	Hermann-Mauguin Symbol	Order of the group	Laue Group
Triclinic	C_1	1	1	$\bar{1}$
	C_i	$\bar{1}$	2	
Monoclinic	C_2	2	2	$2/m$
	C_s	m	2	
	C_{2h}	$2/m$	4	
Orthorhombic	D_2	222	4	mmm
	C_{2v}	$mm2$	4	
	D_{2h}	mmm	8	
Tetragonal	C_4	4	4	$4/m$
	S_4	$\bar{4}$	4	
	C_{4h}	$4/m$	8	
	D_4	422	8	$4/m\ mm$
	C_{4v}	$4mm$	8	
Trigonal	D_{2d}	$\bar{4}2m$	8	
	D_{4h}	$4/m\ mm$	16	
	C_3	$\bar{3}$	3	$\bar{3}$
	C_{3i}	3	6	
Hexagonal	D_3	32	6	$\bar{3}m$
	C_{3v}	$3m$	6	
	D_{3d}	$\bar{3}m$	12	
	C_6	6	6	$6/m$
	C_{3h}	$\bar{6}$	6	
	C_{6h}	$6/m$	12	
	D_6	622	12	$6/m\ mm$
Cubic	C_{6v}	$6mm$	12	
	D_{3h}	$\bar{6}m2$	12	
	D_{6h}	$6/m\ mm$	24	
	T	23	12	$m\bar{3}$
	T_h	$m\bar{3}$	24	
	O	432	24	$m\bar{3}m$
	T_d	$4\bar{3}m$	24	
O_h	$m\bar{3}m$	48		

32 crystallographic point groups in 3d

- (1) Each component in the name refers to a different direction. For example, the symbol for the orthorhombic group, 222, refers to the symmetry around the x , y , and z axes, respectively.
- (2) The position of the symbol m indicates the direction perpendicular to the mirror plane.
- (3) Fractional symbols mean that the axes of the operators in the numerator and denominator are parallel. For example, $2/m$ means that there is a mirror plane perpendicular to a rotation diad.
- (4) For the orthorhombic system, the three symbols refer to the three mutually perpendicular x , y , and z axes, in that order.
- (5) All tetragonal groups have a 4 or $\bar{4}$ rotation axis in the z -direction and this is listed first. The second component refers to the symmetry around the mutually perpendicular x and y axes and the third component refers to the directions in the x - y plane that bisect the x and y axes.
- (6) In the trigonal systems (which always have a 3 or $\bar{3}$ axis first) and hexagonal systems (which always have a 6 or $\bar{6}$ axis first), the second symbol describes the symmetry around the equivalent directions (either 120° or 60° apart) in the plane perpendicular to the 3, $\bar{3}$, 6, or $\bar{6}$ axis.
- (7) A third component in the hexagonal system refers to directions that bisect the angles between the axes specified by the second symbol.
- (8) If there is a 3 in the second position, it is a cubic point group. The 3 refers to rotation triads along the four body diagonals of the cube. The first symbol refers to the cube axis and the third to the face diagonals.

Figure by MIT OpenCourseWare.

Crystal Structure = Lattice + Basis

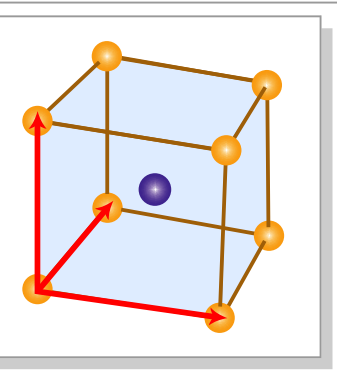
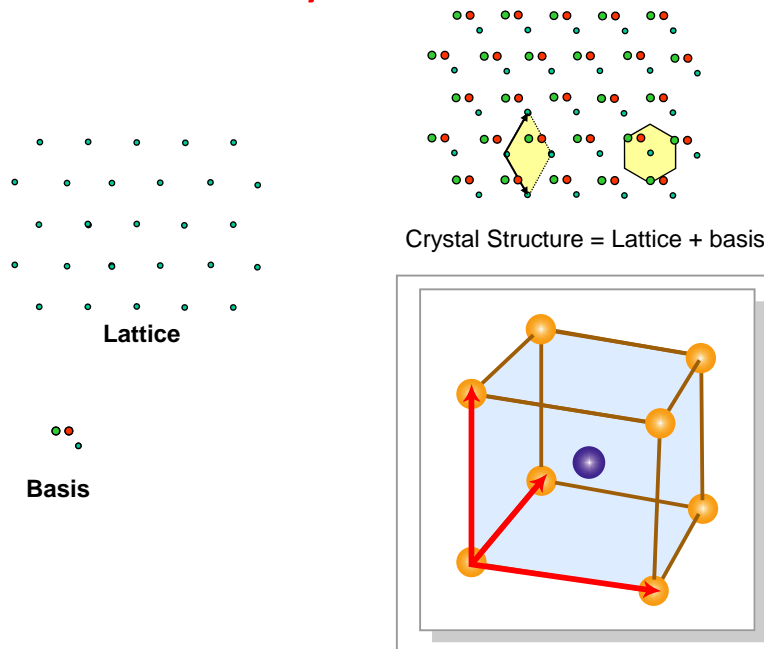


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Primitive unit cell and conventional unit cell

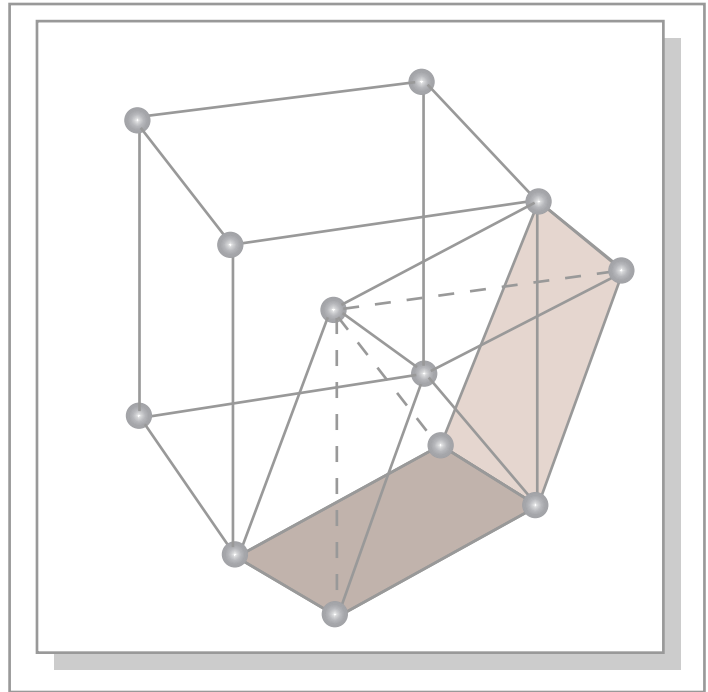
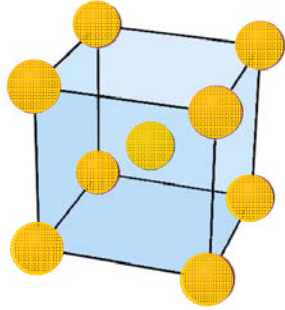
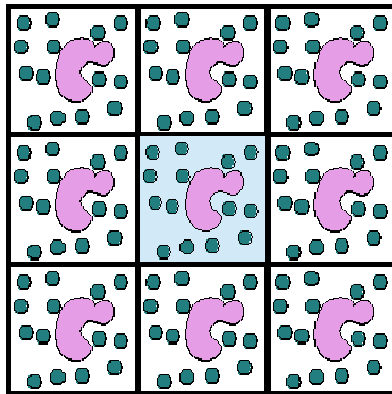


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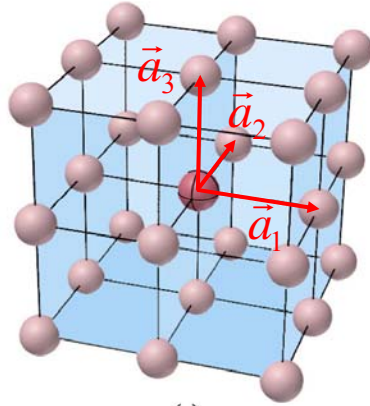
Periodic boundary conditions for the ions (i.e. the ext. potential)



- Unit cell = Bravais lattice = space filler
- Atoms in the unit cell + infinite periodic replicas

Reciprocal lattice (I)

- Let's start with a Bravais lattice, defined in terms of its **primitive lattice vectors**...



$$\vec{R} = l\vec{a}_1 + m\vec{a}_2 + n\vec{a}_3$$

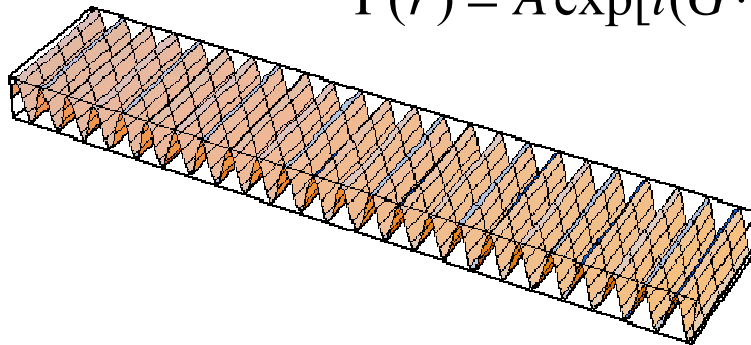
l, m, n integer numbers

$$\vec{R} = (l, m, n)$$

Reciprocal lattice (II)

- ...and then let's take a plane wave

$$\Psi(\vec{r}) = A \exp[i(\vec{G} \cdot \vec{r})]$$



Reciprocal lattice (III)

- What are the wavevectors for which our plane wave has the same amplitude at all lattice points ?

$$\exp[i(\vec{G} \cdot \vec{r})] = \exp[i(\vec{G} \cdot (\vec{r} + \vec{R}))]$$

$$\exp[i(\vec{G} \cdot \vec{R})] = 1$$

$$\exp[i(\vec{G} \cdot (l\vec{a}_1 + m\vec{a}_2 + n\vec{a}_3))] = 1$$

\vec{a}_1, \vec{a}_2 and \vec{a}_3 define the primitive unit cell

$$\vec{G}_i \cdot \vec{a}_j = 2\pi\delta_{ij}$$

\vec{G}_1, \vec{G}_2 and \vec{G}_3 define the reciprocal space Brillouin Zone

Reciprocal lattice (IV)

$$\vec{G}_i \cdot \vec{a}_j = 2\pi\delta_{ij} \quad n \text{ integer is satisfied by}$$

$$\vec{G} = h\vec{b}_1 + i\vec{b}_2 + j\vec{b}_3 \quad \text{with } h, i, j \text{ integers,}$$

$$\text{provided } \vec{b}_1 = 2\pi \frac{\vec{a}_2 \times \vec{a}_3}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)} \quad \vec{b}_2 = 2\pi \frac{\vec{a}_3 \times \vec{a}_1}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)} \quad \vec{b}_3 = 2\pi \frac{\vec{a}_1 \times \vec{a}_2}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)}$$

$$\vec{G} = (h, i, j) \text{ are the reciprocal-lattice vectors}$$

Examples of reciprocal lattices

Direct lattice	Reciprocal lattice
Simple cubic	Simple cubic
FCC	BCC
BCC	FCC
Orthorhombic	Orthorhombic

$$\vec{b}_1 = 2\pi \frac{\vec{a}_2 \times \vec{a}_3}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)}$$